# Al-Ni-Pt (Aluminum-Nickel-Platinum)

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The previous study of this system by [1977Jac] presented a schematic isothermal section at 1060 °C. Recently, two isothermal sections have been determined at 1150 °C by [2005Hay] and at 1100 °C by [2004Gle], respectively.

## **Binary Systems**

The Al-Ni phase diagram [1993Oka] shows five intermediate phases: NiAl<sub>3</sub> ( $D0_{11}$ , Fe<sub>3</sub>C-type orthorhombic), Ni<sub>2</sub>Al<sub>3</sub> ( $D5_{13}$ -type hexagonal), NiAl (B2, CsCl-type cubic, also denoted  $\beta$ ), Ni<sub>5</sub>Al<sub>3</sub> (Ga<sub>3</sub>Pt<sub>5</sub>-type orthorhombic), and



Fig. 1 Al-Ni-Pt lattice parameters of phases along the Ni $_3$ Al-Pt $_3$ Al join [2003Mei]

Ni<sub>3</sub>Al ( $L1_2$ , AuCu<sub>3</sub>-type cubic; also denoted  $\gamma'$ ). The Al-Pt phase diagram [1986Mca] depicts nine intermetallic phases: Pt<sub>5</sub>Al<sub>21</sub> (cubic), Pt<sub>8</sub>Al<sub>21</sub> (tetragonal), PtAl<sub>2</sub> (C1, CaF<sub>2</sub>-type cubic), Pt<sub>2</sub>Al<sub>3</sub> (hexagonal), PtAl (B20, FeSi-type cubic), β (52 to 56 at.% Pt; B2-type cubic), Pt<sub>5</sub>Al<sub>3</sub> (Ge<sub>3</sub>Rh<sub>5</sub>-type orthorhombic), Pt<sub>2</sub>Al (PbCl<sub>2</sub>-type orthorhombic above 1060 °C and Pt<sub>2</sub>Ga-type orthorhombic below 1060 °C), and Pt<sub>3</sub>Al (L1<sub>2</sub>, AuCu<sub>3</sub>-type cubic and low-temperature Pt<sub>3</sub>Gatype tetragonal). The mode and temperature of transition in Pt<sub>3</sub>Al have not been established. [1987Oya] found that the low-temperature Pt<sub>3</sub>Al forms congruently from the AuCu<sub>3</sub>type at ~250 °C through a martensitic transition. The Ni-Pt phase diagram [1989Nas] depicts a large region of the continuous face-centered cubic (fcc) solid solution of Ni and Pt. At low temperatures, two ordered phases appear NiPt ( $L1_0$ , AuCu-type tetragonal) and Ni<sub>3</sub>Pt (L1<sub>2</sub>, AuCu<sub>3</sub>-type cubic), with the critical temperatures at 645 and 580 °C, respectively.

## **Ternary Isothermal Sections**

In the schematic isothermal section drawn by [1977Jac], no ternary phase is present. [1994Kam] arc melted about ten alloys along the Ni<sub>3</sub>Al-Pt<sub>3</sub>Al join, using starting metals of 99.99% purity. The alloys were annealed at 1100 °C for 7 d and air cooled. Structural examination with x-ray powder diffraction showed that the Ni<sub>3</sub>Al-based  $L1_2$  dissolves up to 25 at.% Pt. At 37 and 50 at.% Pt, a ternary phase with the  $L1_0$ , AuCu-type tetragonal structure is present. [1994Kam] denoted this ternary phase as "NiPt<sub>2</sub>Al." In view of its range of homogeneity, this phase is labeled here as  $L1_0$ . Between 26 and 32 at.% Pt, a two-phase mixture of  $L1_2$  and  $L1_0$  is stable. At compositions close to Pt<sub>3</sub>Al, the  $L1_2$  phase is



Fig. 2 Al-Ni-Pt tentative isothermal section at 1150 °C [2005Hay]

### Section II: Phase Diagram Evaluations

again stable. [2003Mei] studied the same join (Ni<sub>3</sub>Al-Pt<sub>3</sub>Al) and found no continuous solution between these phases either in rapid liquid quenching or by heat treatment. The Ni<sub>3</sub>Al-based  $L1_2$  dissolves up to 30 at.% Pt. Between 35 and 50 at.% Pt, the  $L1_0$  phase is stable. The Pt<sub>3</sub>Al-based  $L1_2$ phase is stable between 60 and 70 at.% Pt. At the Pt-end, the low-temperature Pt<sub>3</sub>Ga-type tetragonal phase is stable. The boundaries of the two-phase regions between these phases are not clearly demarcated in the results of [2003Mei]. The lattice parameters determined by [2003Mei] are shown in Fig. 1. The *c/a* ratio of the  $L1_0$  phase is less than unity (0.912 to 0.890), in contrast to the ratio for Pt<sub>3</sub>Al (LT) of more than unity (1.010 to 1.011).

[2004Gle] determined a partial isothermal section for Al-poor alloys of this ternary system applicable for the temperature range of 1150 to 1100 °C. A more detailed study was carried out by [2005Hay] at 1150 °C. With starting metals of ~99.99 wt.% purity, [2005Hay] arc melted alloy compositions for preparing the diffusion couples, which were annealed at 1150 °C for 50 h and quenched. The phase equilibria were determined by a combination of experimental techniques, including high-temperature synchrotron xray diffraction, scanning electron microscopy, and electron probe microanalysis. The partial isothermal section constructed by [2005Hay] at 1150 °C is redrawn in Fig. 2 to agree with the accepted binary data. Except for the ( $\gamma + \gamma'$ +  $L1_0$ ) equilibrium, the triangulations shown in Fig. 2 are schematic.

#### References

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